

Hydrogen – Aspects of safety

Wind Finland 2023



Kiwa Inspecta

Satu Tuurna
4.10.2023

**Trust
Quality
Progress**



Hydrogen

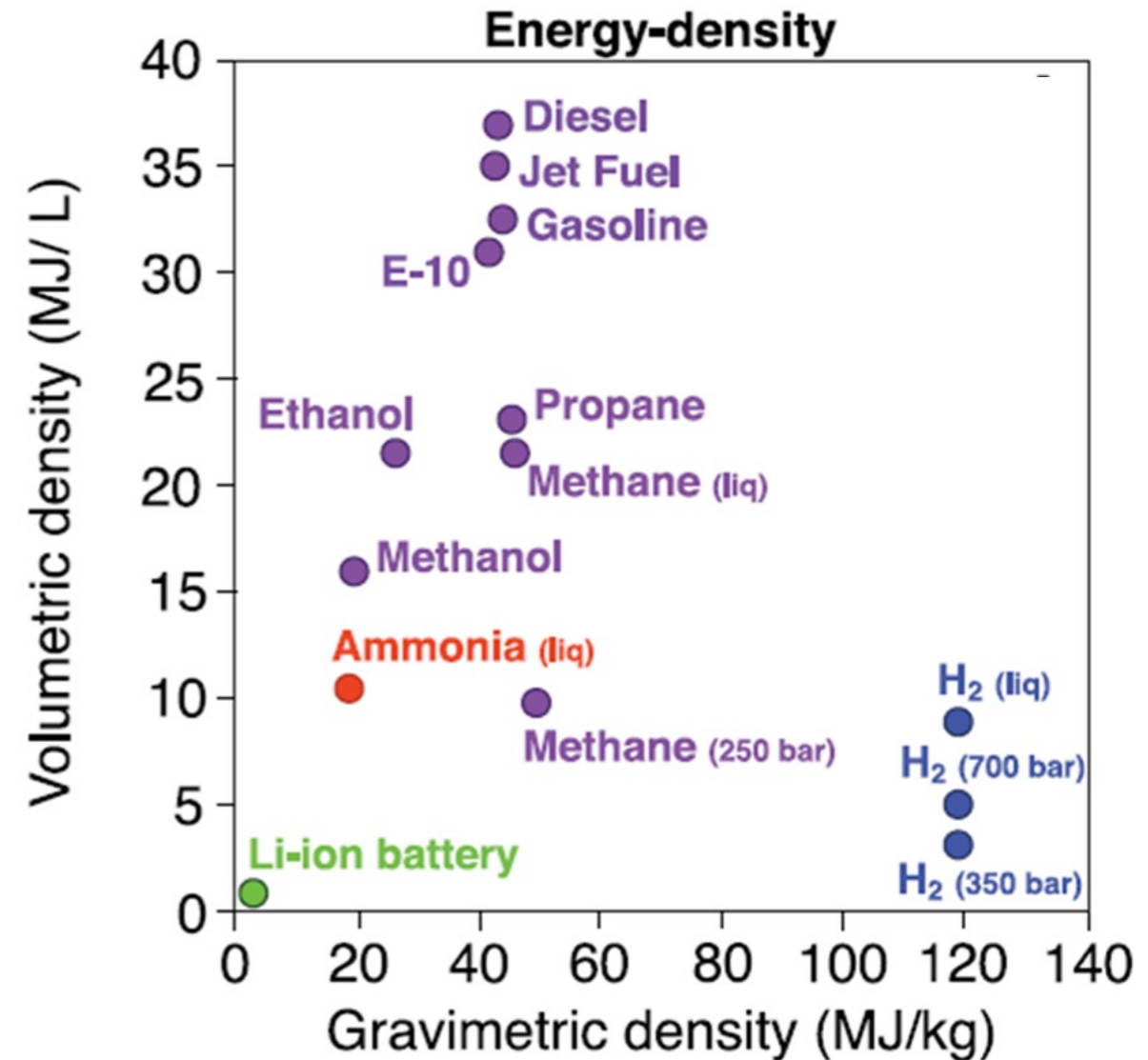
- **Lightest element, at. weight 1 g/mol density 90 g/m³; transparent, no odour**
- **Smallest atom (& molecule, H₂) and common: 75% of the mass of universe**
- **Reactive, on earth rarely free**



- **low density**, diluted in open space, more slowly in closed environment
- **leaks** at joints / contact surfaces and even through walls easier than most gases
- **ignites** at 4-75% concentration, 1/10 of min. Ignition energy of gasoline (0,02 mJ), burns with 2045°C flame to water
- **weakens and embrittles** structural materials

Hydrogen as energy carrier

- Great energy density on a mass basis resulting in minimal weight considerations - lightweight
- Poor energy density on volumetric basis requiring increased space requirements
 - high pressures and low temperatures
- Safety: all flammable/explosive, H₂ easily ignited



Davies et al. Science 2018

Hydrogen – regulatory framework and support

■ EU directives, national regulation

- EU level requirements: **safety**, fair playing field for stakeholders, e.g. REII, SEVESO III, ATEX
- national legislation – **safety**, legal framework
- national licencing and supervisory authorisation on **safety** of hazardous structures and materials/chemicals

■ Standards and guidelines

- international (ISO/IEC, EN, other) standards related to hydrogen
- national, regional, industry/company-specific standards, rules, guidelines, recommendations
- H₂ transition promotes updating

For example: ISO 31010 Risk assessment

prEN 13480-11, ASME B31.12 Piping design

prEN 13445-15 Pressure vessels

EN 16668 Industrial valves

EN 1591-1 Flange tightness (EN 13445-3 Annex G)

EN 13555 Seal performance

Hydrogen embrittlement and cracking of steel

- Hydrogen has a negative effect on the properties of materials, known as **hydrogen embrittlement (HE)**
- Exposure to hydrogen can make some metal vessels or pipelines brittle and increase the risk of cracking, especially where there are pressure variations
- Exposure can lead early, unpredicted failures



<https://uscorrosion.com/index.php/hydrogen-embrittlement-failure-analysis>

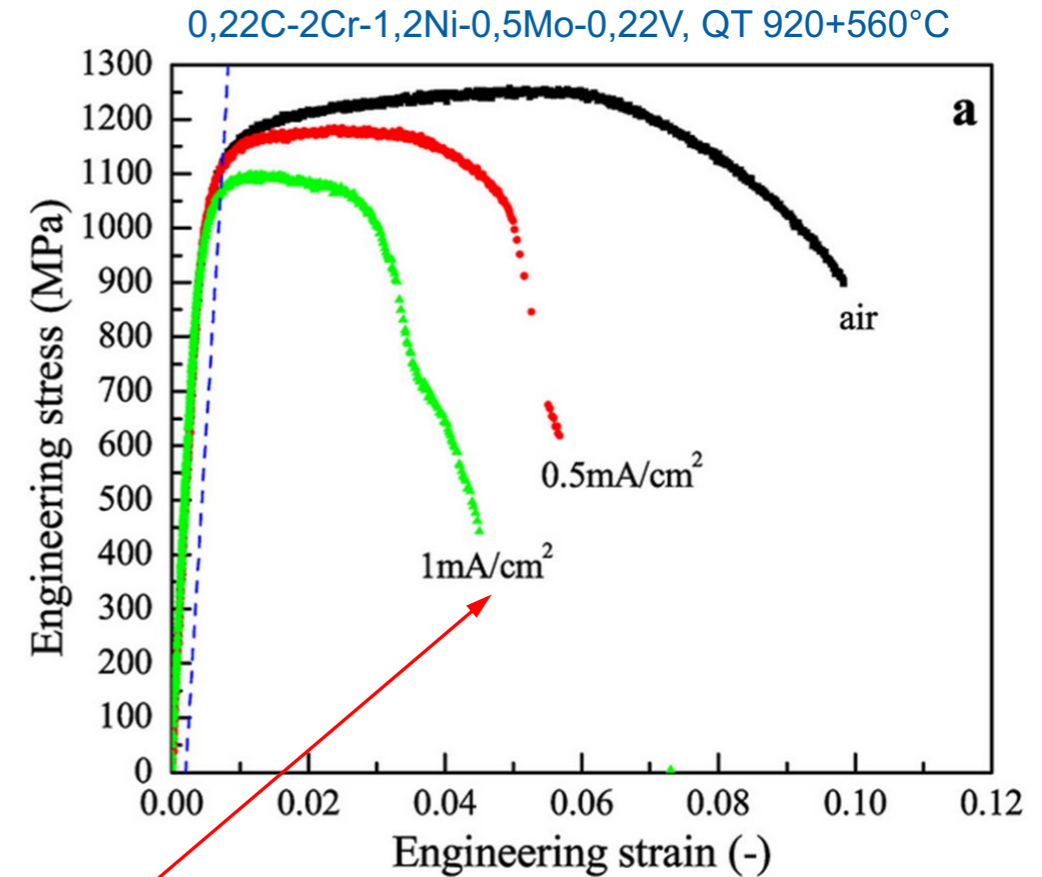
Hydrogen embrittlement (HE) and cracking of steel

■ Factors promoting embrittlement/cracking

- ferritic microstructure, high strength (hardness)
- high pressure, ingress of atomic hydrogen
- fluctuating loading (fatigue)
- construction details are important; manipulations like welding and bending lead to higher strength/hardness → increased HE sensitivity

■ Protective factors

- austenitic microstructure
- hydrogen trapping by microstructural features
- temperature $> 300^{\circ}\text{C}$ → no HE
- ferritic $R_m < 700 \text{ MPa}$, hardness $< 240 \text{ HV}$
→ decreased risks of HE (cycling is still weakening)



Low ductility with hydrogen in steel

Cheng et al. MSEA 2022

Hydrogen – materials selection

Best materials for hydrogen service:

- mostly austenitic steels, e.g. 304, 347, 316L, some Ni alloys

Acceptable: cost-effective ferritic C-, CMn- and Cr-Mo-steels

- limited by allowable strength/hardness, or case-specific acceptance

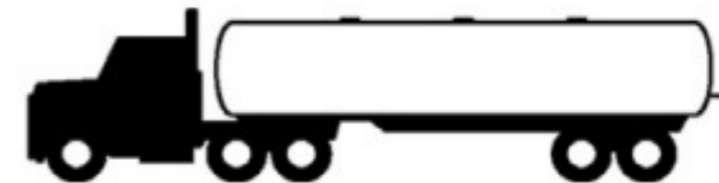
To be avoided in hydrogen environments:

- stainless ferritic, martensitic and duplex steels, cast irons
- Al and its alloys in moist/wet environments, other than oxygen-free Cu
- Ni, Pb, Sn and their alloys; polymers (plastics, with few exceptions)

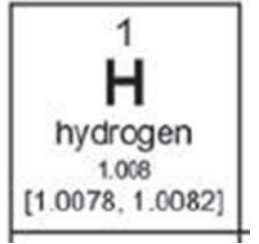


Hydrogen: challenges in operation

- **Industrial production and use**
 - organised, centralised, recorded
 - controlled with planned service & maintenance
- **Local and distributed production and use**
 - expanded exposure to risk
 - variability, including extremes of application
 - potentially reduced control and monitoring
 - need to manage case-specific risks
 - planning, automation, new solutions to compensate for the exposure



Hydrogen – challenges in transport, storage



■ Tube trailers

- in buildings and tunnels: risk of accidents, leaks if insufficiently diluted/ventilated

■ Shipping, pipework, storage vessels

- loading and discharge at harbours
- connections, valves, leak control
- heat transfer for liquified H₂ vessels

■ Issues of **safety** (incl. operation):

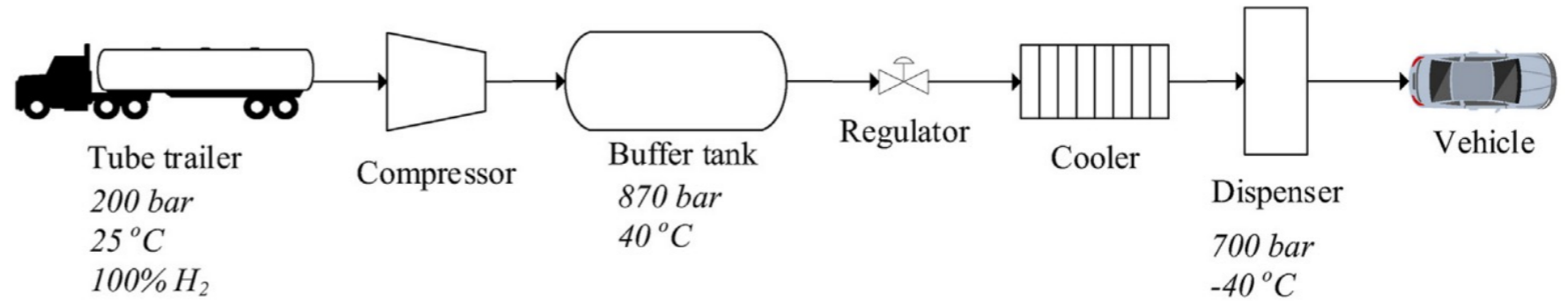
- hazard/risk assessment for facilities
- ventilation, leak & flame detection
- functional condition of equipment
- minimizing risk exposure of personnel
- instructions, checklists for maintenance and operation, training
- completing and signing off work phases
- ... <https://tukes.fi>



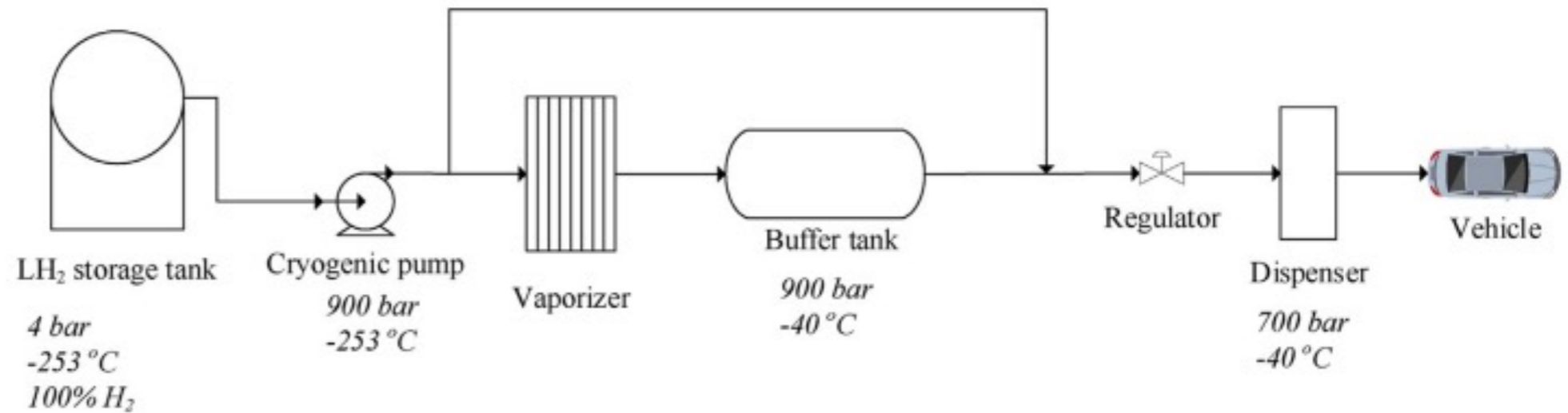
Example: risks of H₂ refueling stations

Yoo et al. Int J Hydrogen Energy 2021

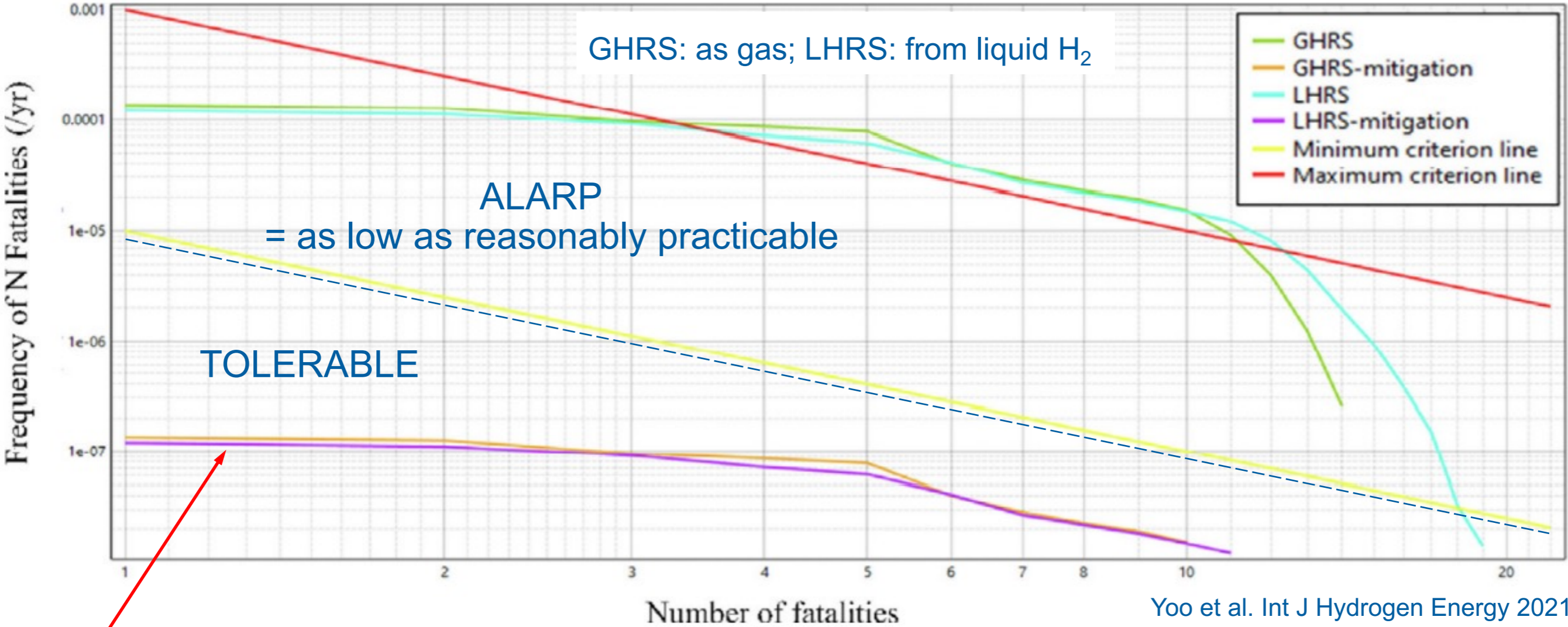
- H₂ as gaseous fuel for fuel cell cars
GHRs
(gaseous H₂ refueling station)



- LH₂ to gas tank and to fuel cell cars
LHRs
(liquid H₂ refueling station)



Risks of H₂ refueling stations without and with added mitigation measures



Yoo et al. Int J Hydrogen Energy 2021

Added measures: safety release, leak detection, automatic and manual emergency stop
→ the assessed risk was reduced by three orders of magnitude to a tolerable level

Summary – opportunities and gaps

■ Low carbon H₂

- rapid expansion of wind, solar, grid connections

■ Gaps and bottlenecks

- EN standards for piping, pressure vessels
- no underground caverns for CO₂ or hydrogen storage
- limited experience on H₂ in non-industrial applications
 - for smaller installations where people can access, it will not be possible to provide the same awareness as in chemical industries
 - **A safe design of the installation is of primary importance**

■ Factors of uncertainty and drivers

- changes of physical and political climate, cost
- developments in systems and technology for safe and cost-effective production, handling and use of hydrogen